organic papers

Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

Akira Mori,* Kanji Kubo, Bing Zhu Yin and Hitoshi Takeshita

Institute of Advanced Material Study, 86, Kyushu University, Kasuga-koen, Kasuga, Fukuoka 816-8580, Japan

Correspondence e-mail: kubo-k@cm.kyushu-u.ac.jp

Key indicators

Single-crystal X-ray study T = 296 KMean $\sigma(\text{C-C}) = 0.005 \text{ Å}$ R factor = 0.052 wR factor = 0.143Data-to-parameter ratio = 16.7

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e. 2-(7-Isopropyl-2-methylcyclohepta[*b*]thiophen-8-ylidene)propanedinitrile

The exocyclic double-bond length of the title compound, $C_{16}H_{14}N_2S$, is 1.366 (4) Å. The planarity of the sevenmembered ring is decreased by the isopropyl group and the thiophene ring. The dicyanomethylene group intersects the seven-membered ring at an angle of 45.9 (2)°. Received 10 January 2001 Accepted 15 January 2001 Online 19 January 2001

Comment

8,8-Dicyanoheptafulvene (DCH), a non-benzenoid aromatic compound, has a brilliant red color and a large dipole moment (7.49 D; Shimanouchi et al., 1966). The introduction of substituent groups to the seven-membered ring causes the color to change. For example, the crystal colors of 1,6-dimethyl-8,8-dicyanoheptafulvene (DMDCH; Shimanouchi et al., 1974a) and 1-isopropyl-8,8-dicyanoheptafulvene (IPDCH; Shimanouchi et al., 1974b) are yellow. It is observed by the Xray crystallographic analyses of DMDCH and IPDCH that the conjugation between the exocyclic double bond and the sevenmembered ring is largely hindered by the substituent at the neighboring position. We now report the structure of the title compound, (I), which has a yellow color, with the aim of contributing to a deeper understanding of the relationships between the 8,8-dicyanoheptafulvene moiety and neighboring substituents.



The seven-membered ring in (I) has a twisted conformation; the intersection angles between the least-squares planes A(defined by C1/C2/C7) and B (defined by C2/C3/C6/C7), and between the least-squares planes *B* and *C* (defined by C3–C6) are 45.2 (3) and 22.2 (2)°, respectively. The deviations of each atom from the seven-membered ring are -0.39 Å for C1, 0.12 Å for C2, 0.22 Å for C3, 0.18 Å for C4, -0.24 Å for C5, -0.18 Å for C6 and 0.22 Å for C7, resulting from the intramolecular steric hindance between the isopropyl and the cyano groups. The dicyanomethylene group is inclined to the seven-membered ring (defined by C1-C7) at an angle of 45.9 (2)°, which is distinct from that (4.8°) of DCH (Shimanouchi et al., 1966), similar to that (47.5°) of DMDCH (Shimanouchi et al., 1974a), and larger than that (39.2°) of IPDCH (Shimanouchi et al., 1974b). These observations result from the intramolecular steric hindrance between the thiophene ring and the cyano groups in addition to the steric

0118 Akira Mori et al. • C₁₆H₁₄N₂S

© 2001 International Union of Crystallography

Printed in Great Britain - all rights reserved



Figure 1

The molecular structure of (I) showing 50% probability displacement ellipsoids.

hindrance between the isopropyl and the cyano groups. The dihedral angle between the thiophene ring and the leastsquares plane B is 22.2 $(1)^{\circ}$. The deviations of each atom from the thiophene ring plane are within 0.02 Å. It is interesting that the exocyclic C1=C8 bond length [1.366 (4) Å] of (I) is shorter than that (1.422 Å) of DCH, and similar to those (1.361 and 1.367 Å) of DMDCH and IPDCH (Shimanouchi et al., 1974a,b).

Experimental

Compound (I) was prepared from 7-isopropyl-2-methylcyclohepta-[b]thiophen-8-one (Ikeda et al., 1994) and malononitrile by reflux in acetic anhydride. Single crystals of (I) were obtained by recrystallization from chloroform.

Crystal data

$C_{16}H_{14}N_2S$	$D_x = 1.247 \text{ Mg m}^{-3}$
$M_r = 266.35$	Mo $K\alpha$ radiation
Monoclinic, $P2_1/a$	Cell parameters from 20
a = 17.174 (5) Å	reflections
b = 7.380(5) Å	$\theta = 10.1 - 18.0^{\circ}$
c = 11.393(5) Å	$\mu = 0.22 \text{ mm}^{-1}$
$\beta = 100.647 (5)^{\circ}$	T = 296 (2) K
V = 1419.1 (12) Å ³	Prism, red
Z = 4	$0.27\times0.27\times0.23$ mm
Data collection	
Enraf–Nonius FR590 diffract-	$R_{\rm int} = 0.039$
ometer	$\theta_{\rm max} = 26.3^{\circ}$
ω –2 θ scans	$h = -21 \rightarrow 21$
Absorption correction: ψ scan	$k = -9 \rightarrow 0$
(North et al., 1968)	$l = -14 \rightarrow 0$
$T_{\min} = 0.959, \ T_{\max} = 1.000$	3 standard reflections
3016 measured reflections	frequency: 120 min
2871 independent reflections	intensity decay: 0.5%
1396 reflections with $I > 2\sigma(I)$	

Refinement

Refinement on F^2	$w = 1/[\sigma^2(F_o^2) + (0.0580P)^2]$
$R[F^2 > 2\sigma(F^2)] = 0.052$	+ 0.0071P]
$wR(F^2) = 0.143$	where $P = (F_o^2 + 2F_c^2)/3$
S = 0.99	$(\Delta/\sigma)_{\rm max} < 0.001$
2871 reflections	$\Delta \rho_{\rm max} = 0.275 \text{ e } \text{\AA}^{-3}$
172 parameters	$\Delta \rho_{\rm min} = -0.287 \text{ e } \text{\AA}^{-3}$
H-atom parameters constrained	

Table 1			
Selected	geometric parameters	(Å,	°).

S1-C11	1.719 (4)	C3-C12	1.422 (5)
S1-C2	1.725 (3)	C3-C4	1.442 (5)
N1-C9	1.137 (4)	C4-C5	1.334 (5)
N2-C10	1.133 (4)	C5-C6	1.436 (5)
C1-C8	1.366 (4)	C6-C7	1.350 (4)
C1-C2	1.441 (4)	C8-C10	1.441 (4)
C1-C7	1.480 (4)	C8-C9	1.441 (5)
C2-C3	1.386 (4)	C11-C12	1.346 (5)
C1-C7-C14-C15	-63.0(4)	C1-C7-C14-C16	172.9 (3)

All H atoms were located at ideal positions and constrained with $U_{\rm iso}$ held fixed to 1.2 or 1.5 (H₂O) times $U_{\rm eq}$ of the parent atoms.

Data collection: CAD-4 Software (Enraf-Nonius, 1989); cell refinement: CAD-4 Software; data reduction: MolEN (Fair, 1990); program(s) used to solve structure: SIR97 (Altomare et al., 1999); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: Xtal_GX (Hall & du Boulay, 1995); software used to prepare material for publication: SHELXL97.

This work was supported by Grant-in-Aid for the Encouragement of Young Scientists (No. 12740381) from the Ministry of Education Science, Sports and Culture, Japan.

References

- Altomare, A., Burla, M. C., Camalli, M., Cascarano, G. L., Giacovazzo, C., Guagliardi, A., Moliterni, A. G. G., Polidori, G. & Spagna, R. (1999). J. Appl. Cryst. 32, 115-119.
- Enraf-Nonius (1989). CAD-4 Software. Version 5.0. Enraf-Nonius, Delft, The Netherlands.
- Fair, C. K. (1990). MolEN. Enraf-Nonius, Delft, The Netherlands.
- Hall, S. R. & du Boulay, D. (1995). Xtal_GX. University of Western Australia, Australia.
- Ikeda, Y. Yin, B. Z., Mori, A. & Takeshita, H. (1994). Eng. Sci. Rep. Kyushu Univ. 16, 295-306
- North, A. C. T., Phillips, D. C. & Mathews, F. S. (1968). Acta Cryst. A24, 351-359.
- Sheldrick, G. M. (1997). SHELXL97. University of Göttingen, Germany.
- Shimanouchi, H., Ashida, T., Sasada, Y. & Kakudo, M. (1966). Bull. Chem. Soc. Jpn, 39, 2322-2331.
- Shimanouchi, H., Sasada, Y., Kabuto, C. & Kitahara, Y. (1974a). Acta Cryst. B30. 1267–1273.
- Shimanouchi, H., Sasada, Y., Kabuto, C. & Kitahara, Y. (1974b). Acta Cryst. B30, 1273-1277.